



## A highly efficient synthesis of (*S*)-(+)-*N*-Boc-coniine using ring-closing olefin metathesis (RCM)

Eunyoung Jo, Youngim Na and Sukbok Chang \*

Department of Chemistry, Ewha Womans University, Seoul 120-750, South Korea

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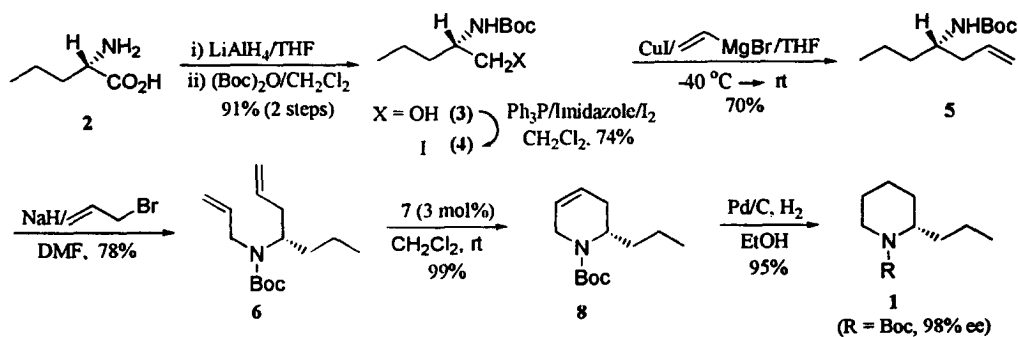
### Abstract

Optically active (*S*)-(+)-coniine as an *N*-Boc protected form was concisely prepared starting from an amino acid, L-norvaline. The key step involved a ring-closing olefin metathesis (RCM) of the dialkenyl compound **6** to give the corresponding cyclic olefin **8** in an essentially quantitative yield. © 1999 Elsevier Science Ltd. All rights reserved.

Alkaloids with a piperidine skeleton are widespread in important natural products. Numerous synthetic strategies for the construction of these physiologically important compounds have therefore been reported.<sup>1</sup> Optically active coniine (**1**, R=H), the poisonous hemlock alkaloid, has served as a building block for many different groups to demonstrate the utility of the synthetic methodologies developed.<sup>2</sup> Herein we report a highly concise synthesis of (*S*)-(+)-*N*-Boc-coniine using a ring-closing olefin metathesis reaction as a key step.

Reduction of a commercially available amino acid (**2**, L-norvaline) followed by *N*-protection in a one-pot operation provided the *N*-*tert*-butoxycarbonyl (*N*-Boc) amino alcohol **3** in 91% overall yield (Scheme 1).<sup>3</sup> The treatment of **3** with I<sub>2</sub> in the presence of Ph<sub>3</sub>P and imidazole transformed the hydroxyl group of **3** to the iodide **4** in 74% yield.<sup>4</sup> The iodide **4** was then homologated by the use of vinylmagnesium bromide (2 equiv.) in combination with copper iodide (1 equiv.) to yield the olefin **5** (70%).<sup>5</sup> The use of either CuBr–SMe<sub>2</sub> or CuCN instead of CuI resulted in slightly lower yields (45–60%). *N*-Allylation of **5** with allyl bromide in DMF provided the dialkenyl compound **6** that served as the precursor for the ring formation reaction. Ring-closing metathesis (RCM)<sup>6</sup> of **6** was performed with the Grubbs' ruthenium benzylidene catalyst Cl<sub>2</sub>(PCy<sub>3</sub>)<sub>2</sub>Ru=CHPh (**7**, 3 mol%) in CH<sub>2</sub>Cl<sub>2</sub> under the atmosphere of nitrogen.<sup>7</sup> The diene **6** was completely consumed within 3 h at room temperature creating the cyclic olefin **8** in an essentially quantitative yield.<sup>8</sup> Palladium catalyzed hydrogenation of the olefin **8** using a hydrogen balloon furnished in 95% yield the *N*-Boc-protected (*S*)-(+)-coniine **1** (R=Boc, [α]<sub>D</sub><sup>20</sup>=+32.8 (*c* 0.43, CHCl<sub>3</sub>)).<sup>9</sup> The spectral data for this compound matched in all aspects those reported in the literature.<sup>2</sup>

\* Corresponding author.



Scheme 1.

In conclusion, we present an efficient synthesis of (*S*)-(+)-*N*-Boc-coniine in seven steps with 35% total yield. If a suitable amino acid was chosen as a starting material, this approach should be amenable to the synthesis of a range of optically active piperidine moieties having a substituent at the 2-position.

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- Representative procedure for the preparation of **8**: To a solution of **6** (0.13 g, 0.51 mmol) in methylene chloride (3 ml) was added the Ru-benzylidene complex **7** (12.6 mg, 0.015 mmol, 3 mol%) and the reaction mixture was stirred at room temperature for 3 h. After removal of the solvent under the reduced pressure, the residue was flash chromatographed on silica gel (ethyl acetate:hexanes=1:20) to afford the cyclic olefin **8** (114 mg, 99%) as a colorless liquid;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 250 MHz)  $\delta$  5.71–5.66 (m, 2H), 4.29 (m, 2H), 3.45 (d,  $J=18.2$  Hz, 1H), 2.43 (d,  $J=17.3$  Hz, 1H), 1.90 (d,  $J=17.3$  Hz, 2H), 1.55–1.49 (m, 2H), 1.48 (s, 9H), 1.11 (m, 4H), 0.91 (t,  $J=6.8$  Hz, 3H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 63 MHz)  $\delta$  155.5, 123.8, 123.3, 79.6, 48.6, 34.1, 28.6, 28.3, 20.1, 14.4; IR (in  $\text{CH}_2\text{Cl}_2$ ,  $\text{cm}^{-1}$ ) 3440, 2962, 2933, 1696, 1601, 1159; HRMS (CI)  $\text{C}_{13}\text{H}_{22}\text{NO}_2$  [ $\text{M}-\text{H}$ ] $^+$  224.1650, found 224.1648.
- The specific optical rotation measured in this work corresponds to 98% ee judging from the known value: Ref. 2a  $[\alpha]_{\text{D}}^{20}=+33.5$  ( $c$  0.43,  $\text{CHCl}_3$ ), Ref. 2b  $[\alpha]_{\text{D}}^{20}=+29.8$  ( $c$  1.3,  $\text{CHCl}_3$ ).